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THE OBJECTIVES OF THIS WORK

1) V. parahaemolyticus, like other non-cholera Vibrio species, contaminates most marine animals in coastal waters, and causes frequently food poisoning (associated gastroenteritis), wound and soft tissue infections, senticenia, and other infections. Among several thousand of infected persons worlwide per year, there are over 10% cases with severe diseases that may end in death when immunocompromized persons are infected. To get more insight into the multidrug resistance (MDR) mechanism of this microbe, particurarly VmrA efflux pump and its function, is one of the objectives:

2) To perform QSAR (Quantitative Structure-Activity Relationship) & chemometric study of structurally unrelated substrates of the VmrA. as extruded by E. coli strains: KAM32 and KAM32/oVCJ6 (with VmrA):

3) To rationalize the results of these studies in terms of molecular features that are responsible for elevated MDR of the VmrA to some of the drugs.

THE VmrA PUMP, ITS MULTIDRUG RESISTANCE (MDR) EFFLUX MECHANISM, AND ITS SUBSTRATES



Proposed structure of the Nar/multidrug transporter VmrA and VmrA-mediated drug efflux from the cytoplasm to periplasm in a Gram-negative cell of V, parahaemolyticus strain AQ334 (M. M. C. Ferreira, R. Kiralj, unpublished results). The VmrA structure was modeled from its primary structure (J. Chen et al., J. Bacteriol., 154 (2002) 572-576.)

Influx of Na⁺ ions in salty modium (coastal water) acuses complex formation of Na⁺ ions with certain Asp and Glu resides in all three domains of the WrnA (transmembrane, cytoplasmic), and periplasmic). This provokes allosteric changes in the structure of this MDR efflux jump, and opens its periplasmic window, the pore, to extrude drug

Independently on the influx of Na⁺ ions, drug molecules are captured by the inner leaflet of the inner membrane and by lateral diffusion or movement are brought to the vestibule. Drug molecules may be captured directly by the cytoplasmic domain and brought to the vestibule, a relatively large window through which they enter into the central cavity where they accumulate before being extruded from the pump.

VmrA is a secondary active transporter that uses electrochemical potential of Na⁺ across the membrane as its energy source. It is a defense mechanism of *V. parahaemolyticus* against several structurally unrelated drugs and xenoblotics.



et al., J. Bacteriol., <u>184</u> (2002) 572-576.] (Table 1). The active organic parts of these twelve substrates (neutral pH) belong to ve distinct classes of organic compounds

Table 1	. Ex	perimenta	l data	for	drugs	1	12

).	CSD source ^a	Formula ^b	pMIC(pVCJ6)*	pMIC(KAM)d	pMIC∆ ^e
	ASUMEG	C14H13N3	3.242	5.961	2.719
	DURDAV01	[C ₂₄ H ₂₀ P]*	3.468	4.672	1.204
r	-	[C ₁₄ H ₁₄ N ₁] ⁺	3.909	5.114	1.205
	ETHIDB	[C ₂₁ H ₂₆ N ₃]*	4.392	4.994	0.602
	CLMPCL02	[C11H12Cl2N2O5]*	5.810	5.810	0
	XAYGEJ	C16H18FN3O3	7.027	7.027	0
	QIMMEE	[C ₂₈ H ₃₁ N ₂ O ₃]*	4.777	4.777	0
	TETCYH10	[C ₂₂ H ₂₅ N ₁ O ₃]*	5.949	5.949	0
	NAVTEJ	[C ₃₇ H ₆₈ O ₁₃ N]*	5.264	5.264	0
)	STOSEH10	[C ₂₁ H ₄₁ O ₁₂ N ₇]*	5.464	5.464	0
L.	GOLWIN	[C ₂₄ H ₃₉ O ₄]	5.318	5.318	0
2	SATLUU	[C ₁₂ H ₁₅ O ₄ S] ⁻	6.461	6.461	0

-concepted to the structure erretrieved from the CSD database. Complete or sparal a database is not complete and the complete or sparal a database is the structure of the complete or sparal and the complete

CBb, and was modeled by Than program. The modeling and biological activity data for drugs 1-12. Appropriate structures from the Cambridge Structural Database (CSD) were retrieved and modified, then optimized at PM3 semi-empirical level. The activities are pMIC = -log(MIC/mol⁺), where MIC is Minima Inhibitory Concentration of the drugs as extruded by two strains of E. coli: KAM32 strain without VmrA, and KAM3239/CJ6 with VmrA (VmrA from V. parahaemolyticus A3334). The pMICa is a measure of the MIDR effect of VmrA.

MOLECULAR DESCRIPTORS

		OKTANT MC			OKO	al delar
Tabl	le 2. Molec	ular descriptors abo	ve the cut-off (0.5	00) in correlati	on with the a	Assisted
No.	Symbol		Definition		R[KAM]* 1	K[pVC36]
8	ly	2nd principal more	ent of inertia		-0.262	-0.621
20	LZ EFE	Sed principal meen los(Dint 1). Din	is molecular di	man also	-0.216	0.721
10		(Titan)	is motecutar of	pore moment	0.404	0.721
14	Nh	No. hydrophobic c	arbon atoms		-0.601	-0.200
15	Na	No. aromatic carbo	in atoms		-0.469	0.645
17	Np	No. polar (not hydr	rophobic) atoms (r	een-H)	0.203	0.517
26	Nr	No, ring atoms (no	n-10		-0.570	-0.410
27	wh	NhNt, number fra	ction of hydropho	bic atoms; Nt	-0.566	-0.726
		is total No. non-H	atoms		0.107	
50	we	Nh/Nt rareber fra	ction of bodroom	honding non-	0.567	0.579
		Hatome Nh is No	HB donors/accer	dors.	*1048	446.5
30	wp	Np/Nt, number fra	ction of polar aton	m#	0.566	0.726
31	wi	NI/Nt, number fras	tion of planes ato	ens; NI is No.	-0.177	0.553
		non-H atoms in all	planar fragments			
-40	WON	NON'N, number	fraction of O/N at	toms; NON is	0.454	0.558
		No. onygen and nit	trogen atoms			
45	EHR	B/NI, No. bonds p	er atom; B is No	bonds (non-	-0.415	-9,693
4.6		10 Nobit much se fax	tion of size stores		-0.425	0.713
10	DAT	contexting displays	consol (Ed motho	AMORICO	0.42.5	0.799
54	E4	average polarizabil	ity (E4 method A	ROPACE	-0.611	-0.227
55	BT	ß Inpersolarizabi	lity along the di-	pole moment	-0.663	-0.456
		(E4 method, MOP.	AC)	pere memoria		
59	GM.	absolute average	y hyperpolari	ashility (E4	-0.108	-0.508
		method, MOPAC)				
63	Wn3	W(N) ² , normalia	red Wiener inde	x W is H-	0.510	0.221
		depleted Wiener in	dex			
64	BOMO-1	energy of HOMO-	1 orbital		0.584	0.375
65	HOMO	energy of HOMO	stetal		0.588	0.376
68	8t-	the most negative l	257 alorne charge	r (non-H)	-0.672	-9.675
09	Sec.	the most positive E	or atomic charge	(1001-11)	0.420	0.541
24	a data	(Qr) = (Q-), the In-	ges ESP charge of	midfence	-0.539	-9,636
(0)	-sexps.	steps's, surface any	te statuon or hyd	repainter UH	-0.203	4.568
		independence in the second sec	H atoms and	molecula		
		respectively		. moneole,		
78	Mrefu	Mref/N. molecula	refractivity per a	nour. Mref is	-0.259	0.606
		molecular refr	activity (Clog	P method.		
		CHEM3DK N is N	o. all atoms			
79	Enbt	(HOMO-1)/N, HO	MO-1 orbital ener	gy per atom	0.233	0.538
80	Enb	(HOMO)/N, HOM	O orbital energy p	er stom	0.232	0.529
83	Enb-Hi	RHOMO-DUHO?	(O) N. liontier (orbital energy	0.232	0.534
		sum (HOMO-1)+4	HOMO) per atom			
81	Enb-II	[IIIOMO-I) (LU)	40)]/N. frontier e	rbital energy	0.328	0.501
		sum (HOMO-1)+4	LUMO) per atom			
87	DMm	DM/N, DM per atc	ana -		0.819	0.788
88	Hin	BTN, hyperpolarit	zability β per aton	٠.	-0.743	-0.551
89	Бра	B:Nt, No. bouids pe	r atom: B is No. b	Jonds	-0.165	0.633
20	L .	No. non-11 stores a	tong the longest b	orad chain	0.215	0.540
25	Anno	nititinum X coord	mate		-0.398	-0.528
100	A FILLING	reaccinging 5 contra	mare		-0.658	-0.138
104	1111	Your You's orals	and an income so in the		0.647	(107.4
1.01	D2	Zuars-Zmin mole	ular box height		-0.538	-0.015
104	sighb	MeSm HB donor	electronic established	e density: Sm	0.496	0.533
		is molecular surfac	e area			
105	sighyd	Nh/Sm. hydrouholi	ic curbons surface	density	0,660	0.752
105	sign	Na/Sm. aronalic c.	arbons surface der	nily	-0.347	0,686
107	sigp	NprSm. polar atom	s surface density		0.494	0.669
108	sigr	Nr/Sni, ring atoms	surface density		-0.426	-0.671
109	sigon	ON/Sna, ON atoms	surface density		0.375	0.536
130	ntb	Nt/B, non-H atoms	per bood		0.469	0.629
111	Np2	(Np - 12H, square)	function of Np		-0.527	-0.787
112	ON2	(ON - 10F, square	function of ON		-0.463	-0.593
113	wb2	twh 0.61 square	function of wh		-0.682	-0.784
114	wu2	two - 0.31 ⁵ , square	function of wa		-0.168	0.627
115	wp2	twp = 0.41 ² , square	function of wp		-0.682	-0.784
116	Mrefn2	(Mrefn - 0.19F, sq	uare function of N	frelio _	-0.485	-0.745
117	sigp2	tsig - 0.02212; sij	r - Nye/Sm, vali	ance electron	-0.609	-0.674
		surface density; No	e is No. valence o	dectrons		
118	RD	ratio of actual and	standard bond long	spin anu.	0.232	0.662
119	RD2	(RD - 0.98). squar	e function of RD		0.583	0.787
120	RRD	1/RD, inverse of R	D		-0.243	-0.6*3
N'or	retation ed	etticsent with netic	ny pMIC(KAM).	"Correlation of	sortheient wi	Its netivity
p.ati	c (ph CO6). * standard 5-*	one routh lengths said	rescalle it atoms,	scentil hout len	gues are trout e	augs 1 - 13 emiles: C d
CH CH	-CH1. C-5	(CH-NHA) C-P (C	R. PH.I. S-O INS-	OBL C-FICH	 -FL C-Q (CH) 	OH: N-C
me	-on, e-c	(CH)-CIL				
		SELECTED N		DESCR	PTORS	
тэ	ore 3. Selet	teu moleculur deser	iptors for drugs 1	• 12		
X	PT.	NR	11	1		36-1-2
20	C PP	ru wr J	arrau. HOMO	signya P	spa wh2	.supoin2
-	0.12.2	11 0.8827	eV	A /	P1 /14600	cm' mol
	0.4236	14 0.8824 1	10.1.649 -1.837	0.0510	81 0.02500	0.20430
2	0.0064	24 0.9600 0	13.003	0.0522 1	0.1296	0.16122
2	0.4007	21 0.8237	11 CIN 11 200	0.0590	91 0.00566	0.16020
- 4	0.2931	10 0.8535 1	antata - 11.251 120.021 - 1.0.251	0.0303	a) 0.0176 1 0.0176	0.17074
2	1.69.60	1.1 0.6937	10.204 -10.204 174.400 -6.244	0.0396	 0.0000 0.0000 	0.015420
	1.06.90	27 0.6061	0.343	0.0390	26 0.0176	0.01262
- 7	V.9090	10 0.5625	19.426 .11.266	5 0.0135	1 0	0.00250
2	11522			2 0.0152		- AM220U
8	1.1533	15 0.5002	11.86	7 0.0166	16 0.0071	0.00519
8 9	1.1533 2.3617 1.1929	35 0.5098 2	316.371 -11.86	7 0.0166	16 0.0071	0.09518
8 9 19	1.1533 2.3617 1.7979 1.5071	35 0.5098 1 20 0.1250 1 23 0.6071	816.371 -11.86 249.845 -13.86 176.495 -4.333	7 0.0166 1 0.0329 0.0526	16 0.0071 61 0.0100 49 0.0490	0.09518 0.07617 0.07894

The descriptors were calculated by Titan, MOPAC and Chem3D using optimized geometry, and from molecular formula.

PLS (PARTIAL LEAST SQUARES) AND PCR (PRINCIPAL COMPONENT REGRESSION) MODELS

Parameter	PLS (pVC36)	FCR (pVC36)	PLS (KAM)	PCR (KAM)
P('s (?a)	1 (20.0)	1 (70.0)	1 (52.8)	1 (55.6)
SEV	0.721	0.683	0.452	0.413
SEP	0.562	0.565	0.368	0.589
Q	0.803	0.818	0.762	0.763
R	0.889	0.888	0.866	0.849
4	0.410	0.410	0.288	0.270
FF	0.206	0.209		
Nh			4).228	-0.260
wr	-0.201	-0.212		
E4			-0.231	-0.261
HOMO			0.225	0.190
sigtryd	-0.215	-0.217	-0.250	-0.258
Np2	-0.225	-0.215		
wh2			4).258	-0.250
Mrein2	-0.213	-0.210		

42. 40,213 40,214 40,214 40,214 40,214 40,214 40,214 40,214 40,214 41 *Paranet that they

shows drawna drawdiet hen expansed a untry value. The PLS & PCR regression statistics for modeling of the both activities shows that there is no significant difference between the PLS and PCR models. The regression coefficients are in favour for pMIC(pKU.Efg.), whils the errors are smaller for pMIC(pKM). The regression vectors agree with the correlation analysis (Table 2) and chemical interpretation of the MDR phenomenon.



Predicted activities, absolute and relative errors for drugs 1-12, Predicted activities, absolute and relative errors for drugs 1-12, and the predicted MDR parameter pMICL There are 4 drugs with relative error 5 ro PMIC(pVCJ8), and only 1 drug with such error for pMIC(KAM). The pMICL parameter shows correctly the presence and absence of elevated multidrug resistance of VmA for 9 from 12 drugs (75%). All regression and other chemometric analyses in this work were performed by using programs Piroutte and Matlab on autoscalled data, and leave-one-out crossvalidation.





The pMICA-descriptor correlation plots that show which molecular features are responsible for distinction of drugs to which WmX has elevated multidrug resistance (1-4) or there is no significant resistance (5-12). These molecular properties account for the content of unsaturated bonds, ring and planar its, aromatic and polar groups



12

The HCA dendograms with samples and complete linkage and using the same molecular descriptors as in previous PCA analyses. Two classes of drugs can be clearly observed in each dendogram, whits sub-clusters may differ from those from the PCA plots. Two-membered sub-clusters reflect the structural similarity of their members.

CONCLUSION: The KAM32 strain prefers rather hydrophobic species. The KAM32/pVCJ6 strain, and consequently the VmrA efflux pump, prefers rather rigid and condensed heteroaromatic species.

ACKNOWLEDGEMENT: FAPESP



The PCA (Principal Component Analysis) scores (top) and loadings (bottom) plots using the 5 descriptors from regression modeling of pMIC(pVCL8). The activity classes can be observed: good (6), moderately good (M) and poor (P1, P2) substrates that be higher content of aromatic and hydrophobic groups, and lower content of plane groups, small dipole moments and molar refractivity.



The PCA scores (top) and loadings (bottom) plots using the 5 descriptors from regression modeling of pMIC(KAM). Two classes of substrates can be observed: E – more elongated, linear or chain-like species; B – more branched, cyclic or spherical species, as substrates of other pumps in *E*. coli KAM32. B species are in average better substrates than E species. Better substrates may be characterized by higher content of hydrophobic groups, more negative HOMO and smaller polarizability.